

Name: _____

Date: _____

Molarity

Complete the table for the following aqueous solutions

	Solute	Mass Solute (g)	Volume (liters)	Molarity mol/L
1.	LiBr	43.3	0.5	_____
2.	NH ₄ Cl	160.2	4.0	_____
3.	KOH	28.0	0.25	_____
4.	NaCl	146.0	25.0	_____
5.	BaCl ₂	385.5	_____	2.5
6.	KCl	37.5	_____	0.125
7.	NaOH	80.0	_____	1.6
8.	CuSO ₄	_____	1.6	0.125
9.	Al(NO ₃) ₃	_____	0.5	0.6
10.	(NH ₄) ₂ SO ₄	_____	0.3	2.0